

# Xenon-Doped Liquid Argon Scintillation Light

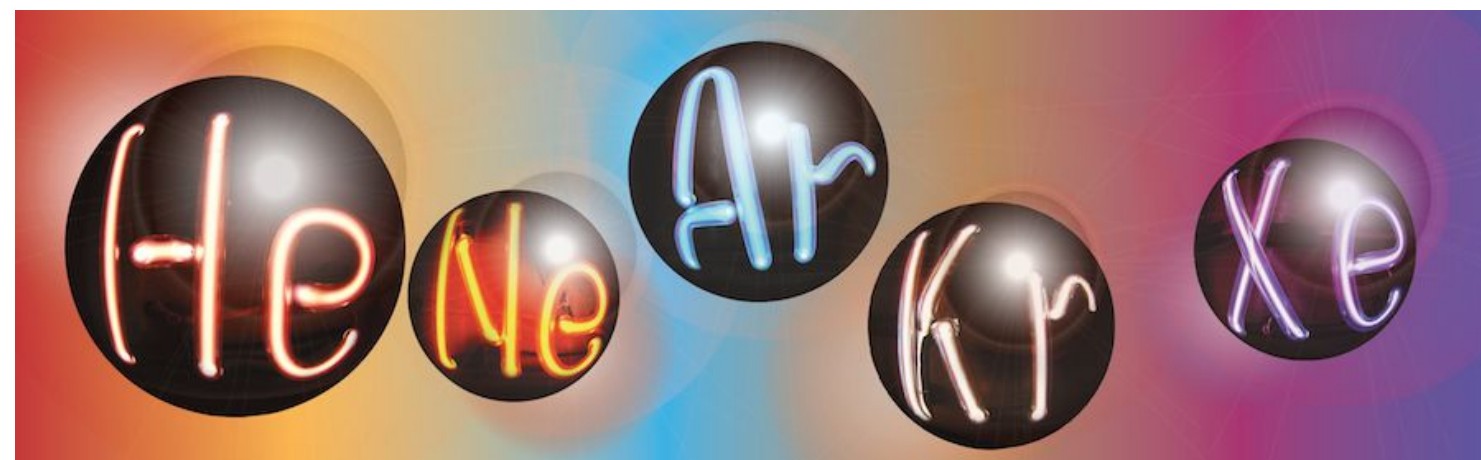


D.E. Fields, M. Gold, T. Reza, Luis Flores-Sanchez (UNM)

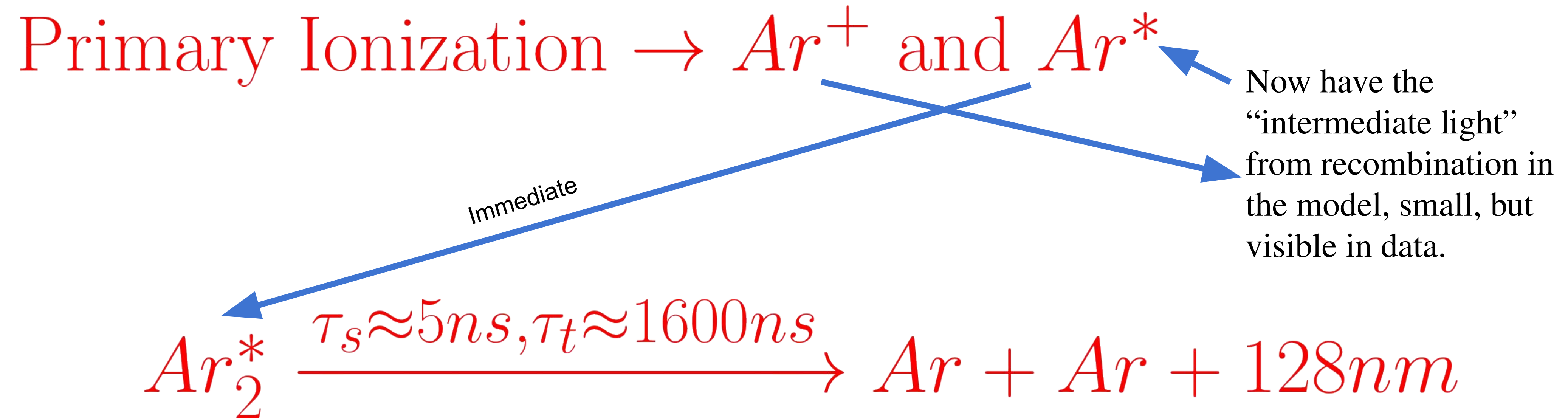


S.R. Elliott, R. Massarczyk, A. Mazumdar, C. Romo Luque, T. Thorpe, N. O'Brien, B. Turner (LANL)

<https://arxiv.org/pdf/2009.10755.pdf>



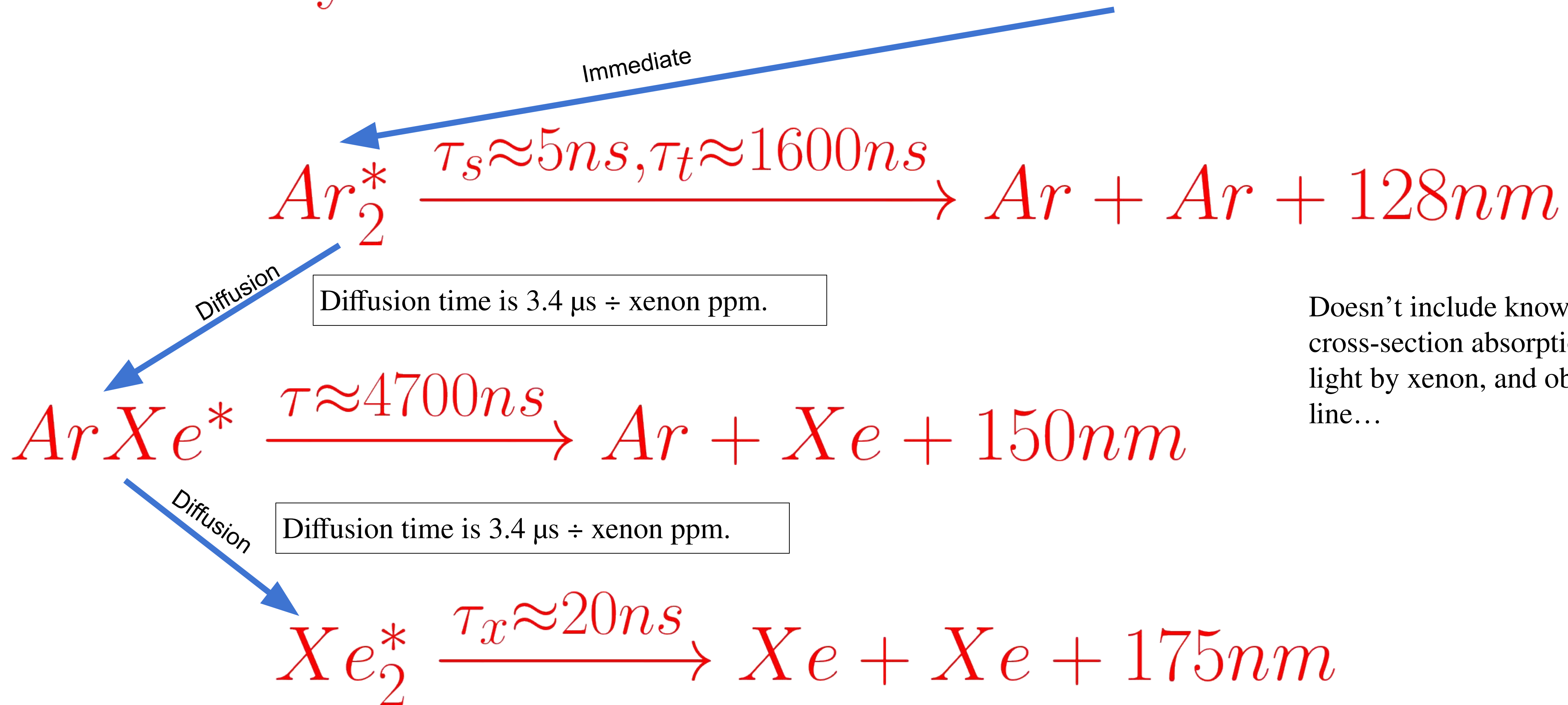
# Evolution of Our Understanding: *Pure* LAr



Very different measurements of the long, triplet lifetime from 1100 to 2100 ns. Ratio of triplet to singlet different for different ionizing particles and can change in the presence of an electric field. Doping with xenon has dramatic effects...

# Evolution of Our Understanding: *Xe-doped* LAr

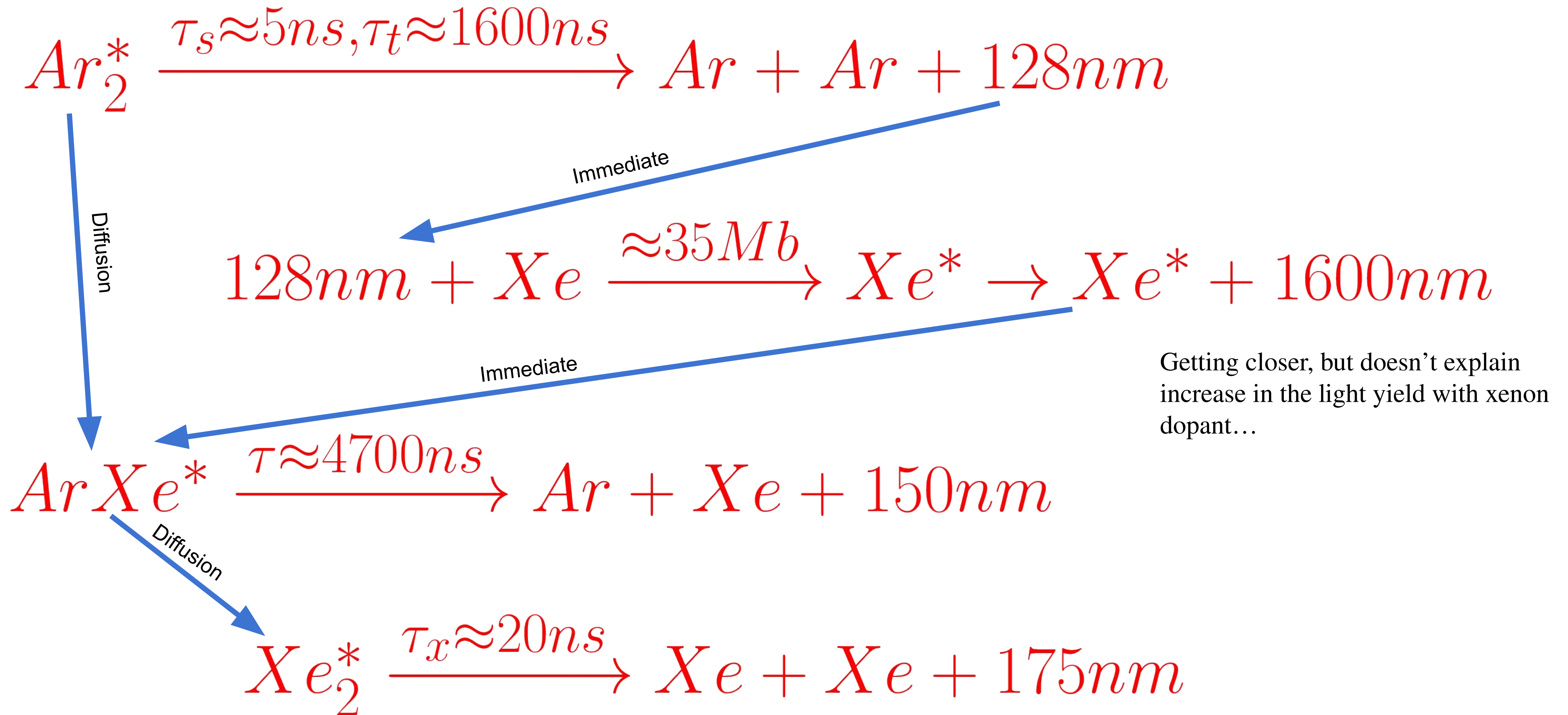
Primary Ionization  $\rightarrow Ar^+$  and  $Ar^*$



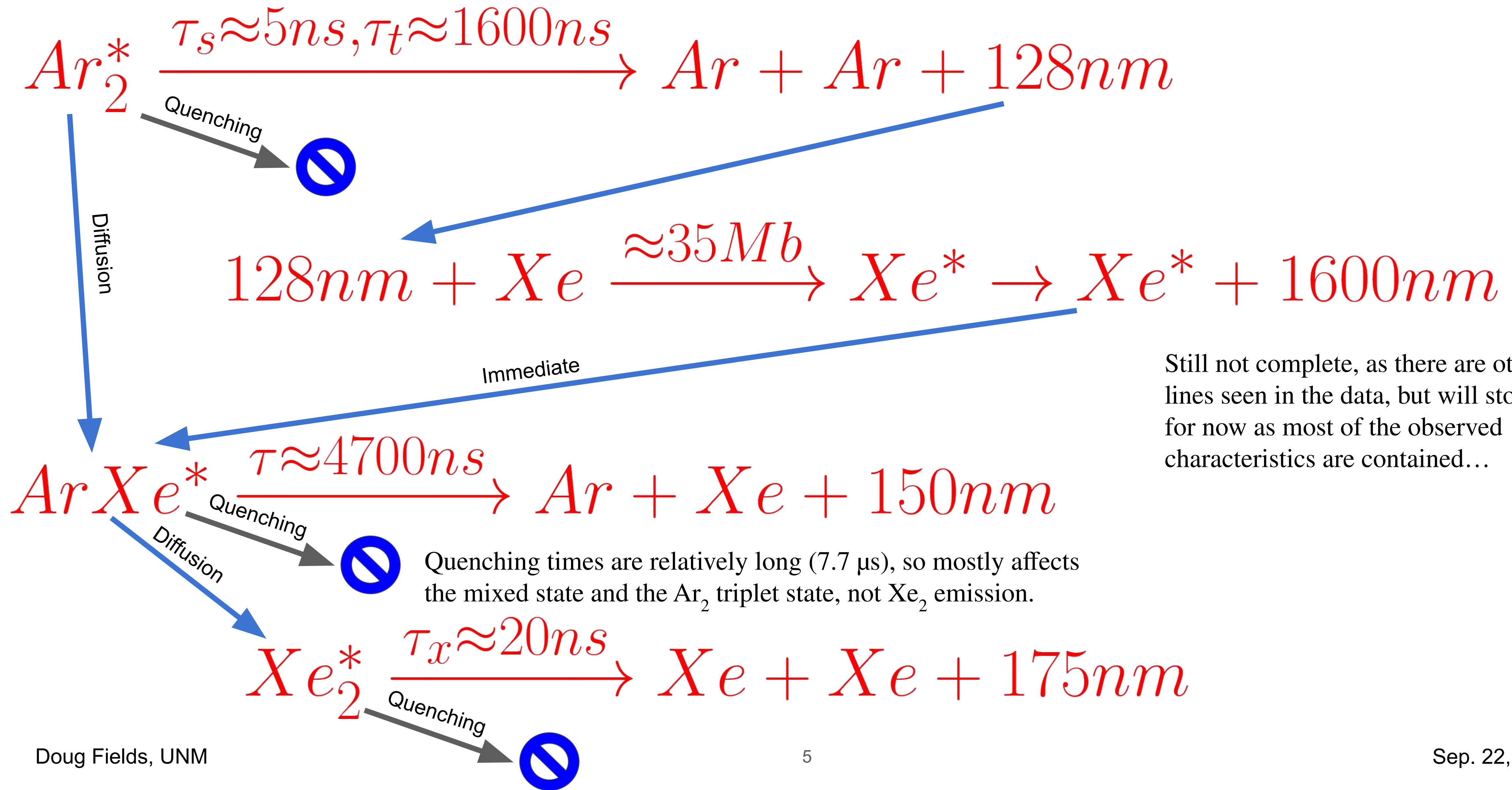
Doesn't include known, large cross-section absorption of the 128 nm light by xenon, and observed 1600 nm IR line...



# Evolution of Knowledge: Absorption



# Evolution of Knowledge: Quenching



Still not complete, as there are other IR lines seen in the data, but will stop here for now as most of the observed characteristics are contained...

# Absorption is not straightforward

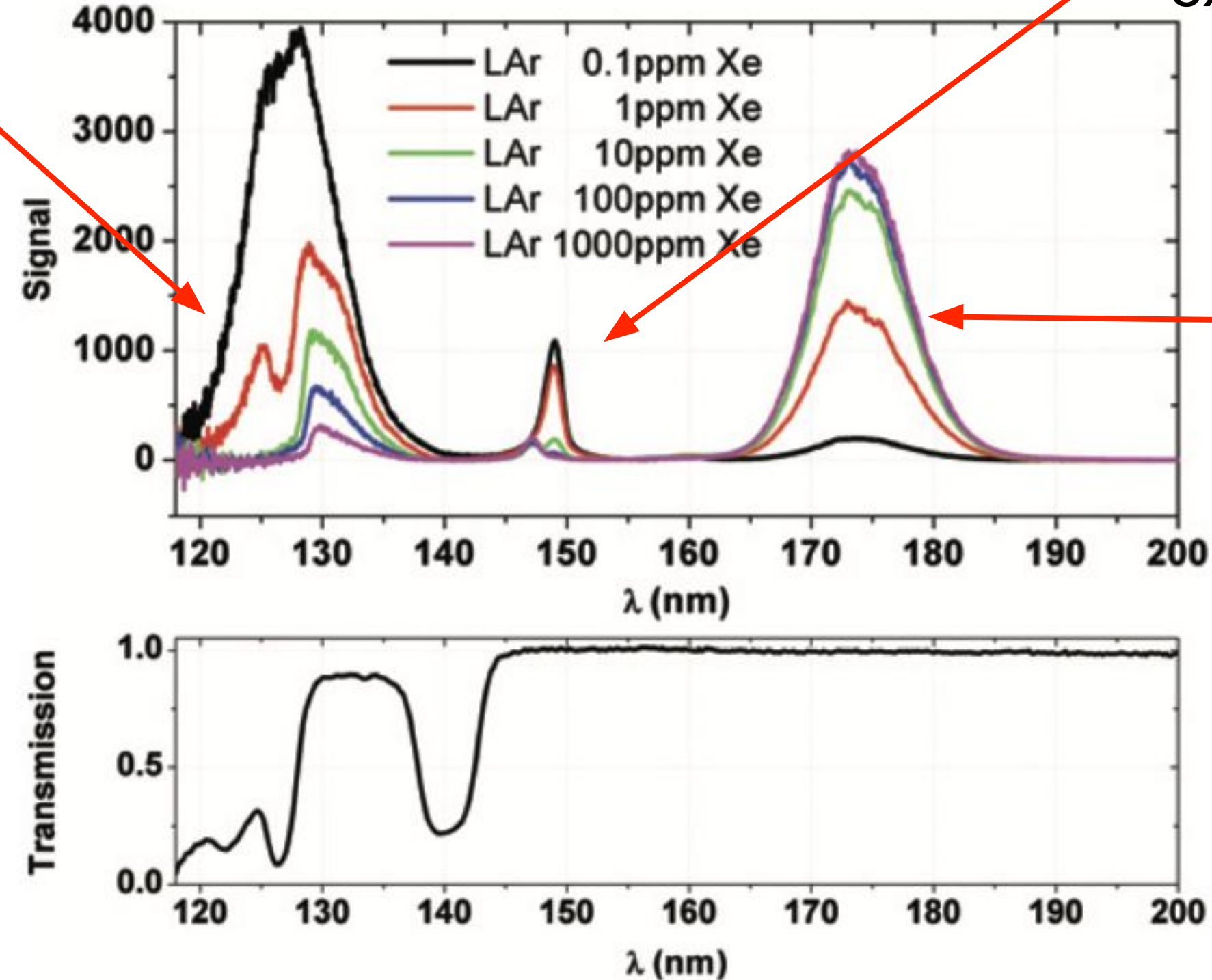
A. Neumeier *et al.*

The VUV emission of electron-beam excited liquid argon doped (2 mm)

Argon  
excimer

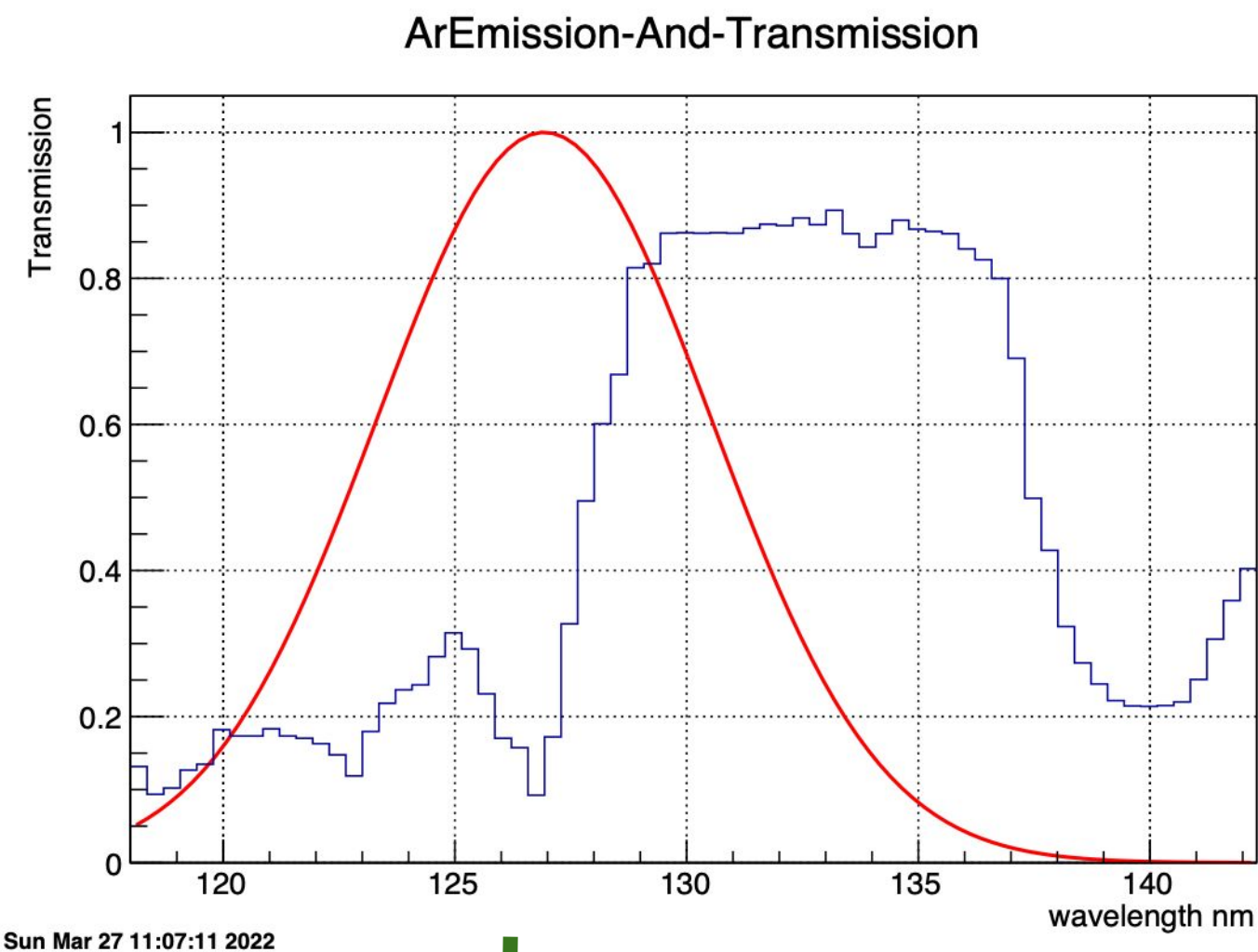
Mixed (ArXe)  
excimer

Xenon  
excimer

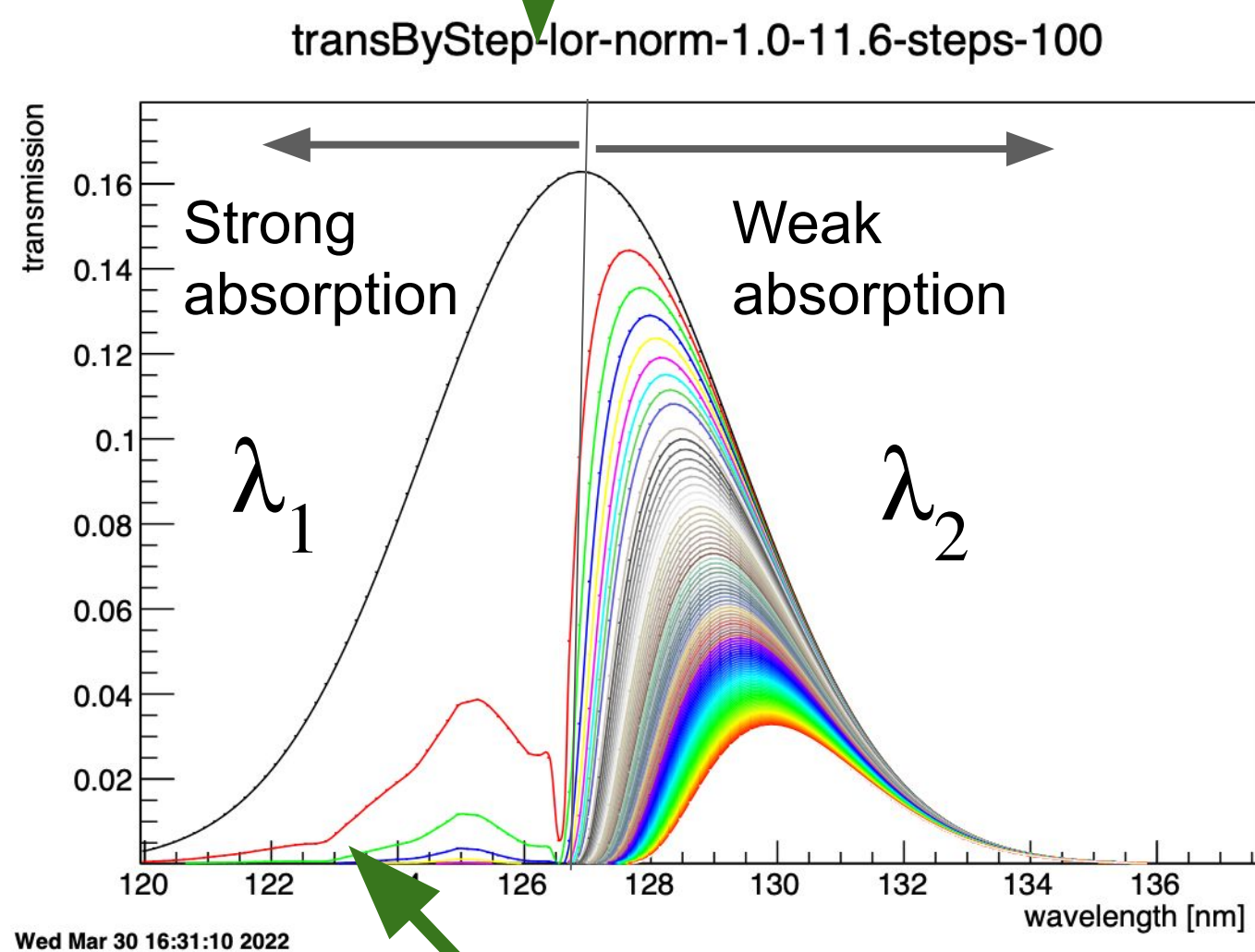


Transmission of liquid argon doped with 0.1 ppm xenon measured with a deuterium light source and a length of the optical path of 11.6 cm





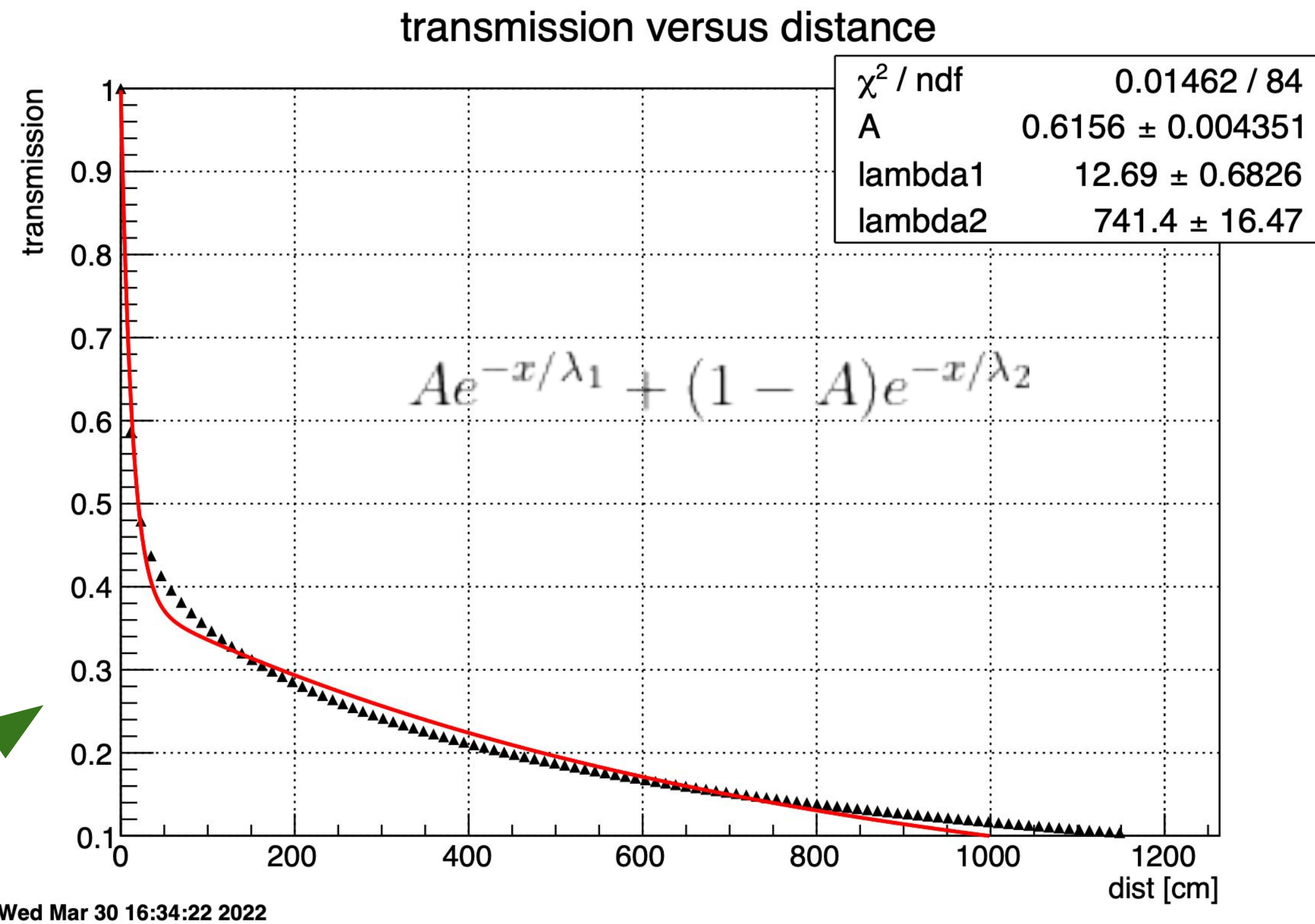
Iterate convolution



Red curve is 11.6 cm (0.1 ppm Xe),  
green is 23.2 cm, etc.

∫

Integral



For  $\lambda_1 = 12.7$  cm,

$$\sigma_{\text{abs}} = 47.6 / (12.7 \text{ cm}) / (0.1 \text{ ppm}) = 37.5 \text{ Mb}$$

in agreement with Calvo *et al.* ArDM

# Our Model

time dependencies of the different states. Our model consists of the following coupled differential equations for the number of molecules as a function of time  $t$  for the dimer states argon singlet  $S(t)$ , argon triplet  $T(t)$ , mixed  $M(t)$  and xenon  $X(t)$ :

$k_x$  is calculated diffusion limited reaction rate in LAr for Ar, to Xe (scales with Xe concentration)  $k_x = 2.9 \times 10^{-4} \times [\text{ppm}] \text{ ns}^{-1}$

$$\dot{S} = -S/\tau_S - (k_x + k_q) S \equiv -\lambda_1 S \quad (1)$$

$$\dot{T} = -T/\tau_T - (k_x + k_q) T \equiv -\lambda_3 T \quad (2)$$

$$\dot{M} = - (1/\tau_M + k_x + k'_q) M + (k_x + A/\tau_S) S + (k_x + A/\tau_T) T \quad (3)$$

$$\dot{X} = -X/\tau_x + k_x M \quad (4)$$

Note: We are adding the intermediate light in the model (can see the effect).

We take A from the fit to the absorption curve.

Note: Below 0.5 ppm, quenching is faster than diffusion.

$k'_q, \sim k_q = 1.3 \times 10^{-4} \text{ ns}^{-1}$  are quenching rates (Segreto, *Phys.Rev.D* 103 (2021) 4, 043001 )



# Light yield (time)

Then these equations are easily integrated giving  $S(t)$ ,  $T(t)$ ,  $M(t)$  and  $X(t)$ , and the light seen (128 nm + 150 nm + 175 nm) is given by:

$$l(t) = N_1 (1 - A) e^{-t\lambda_1} / \tau_S + N_3 (1 - A) e^{-t\lambda_3} / \tau_T + M(t) / \tau_M + X(t) / \tau_X$$

In terms of singlet fraction (sfrac)  $N_1 = N * \text{sfrac}$ ,  $N_3 = N(1 - \text{sfrac})$

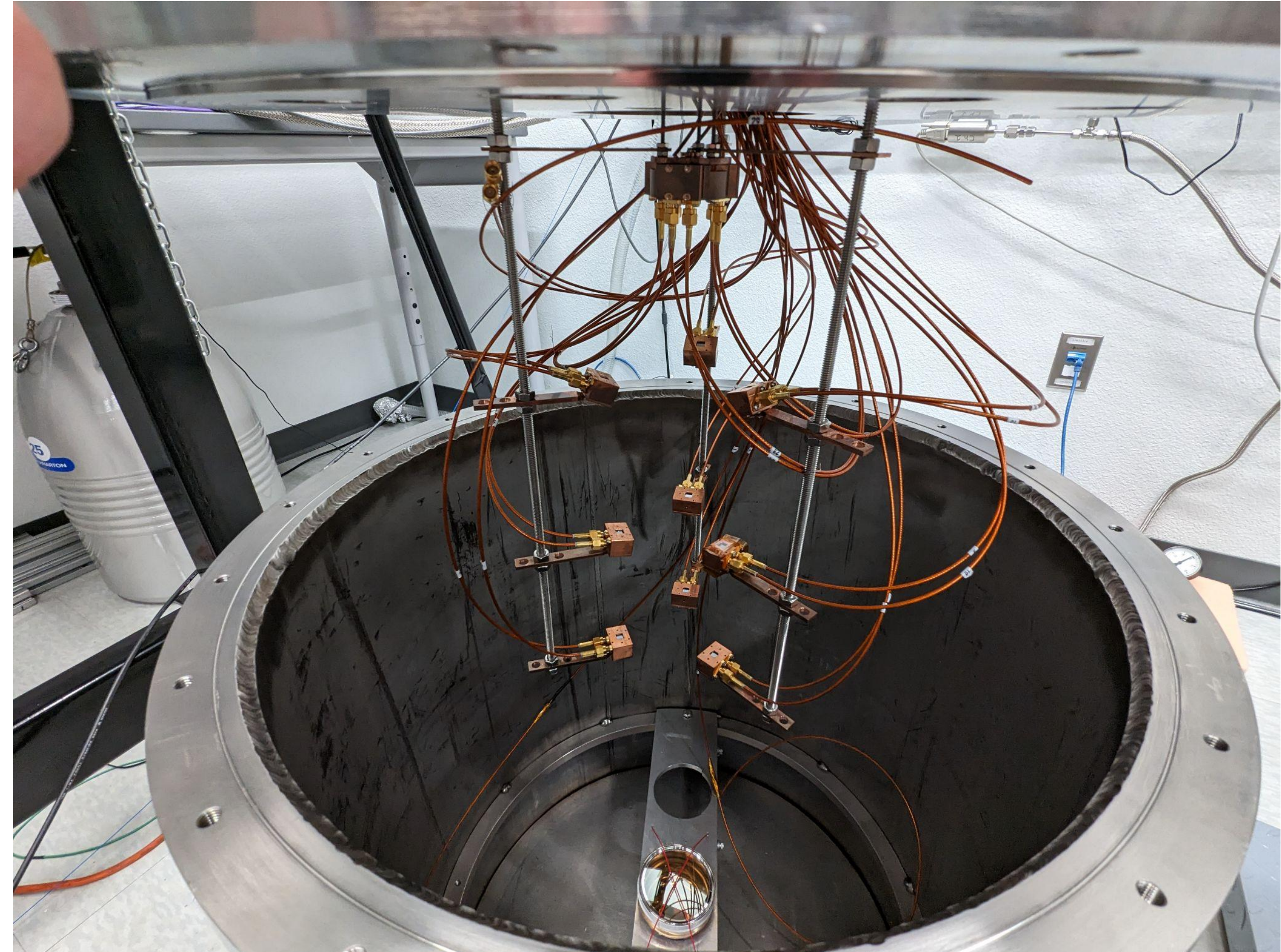
$N$  is the total number of dimers created by initial ionization

Similar to Segretto, but includes absorption, and is based on diffusion times



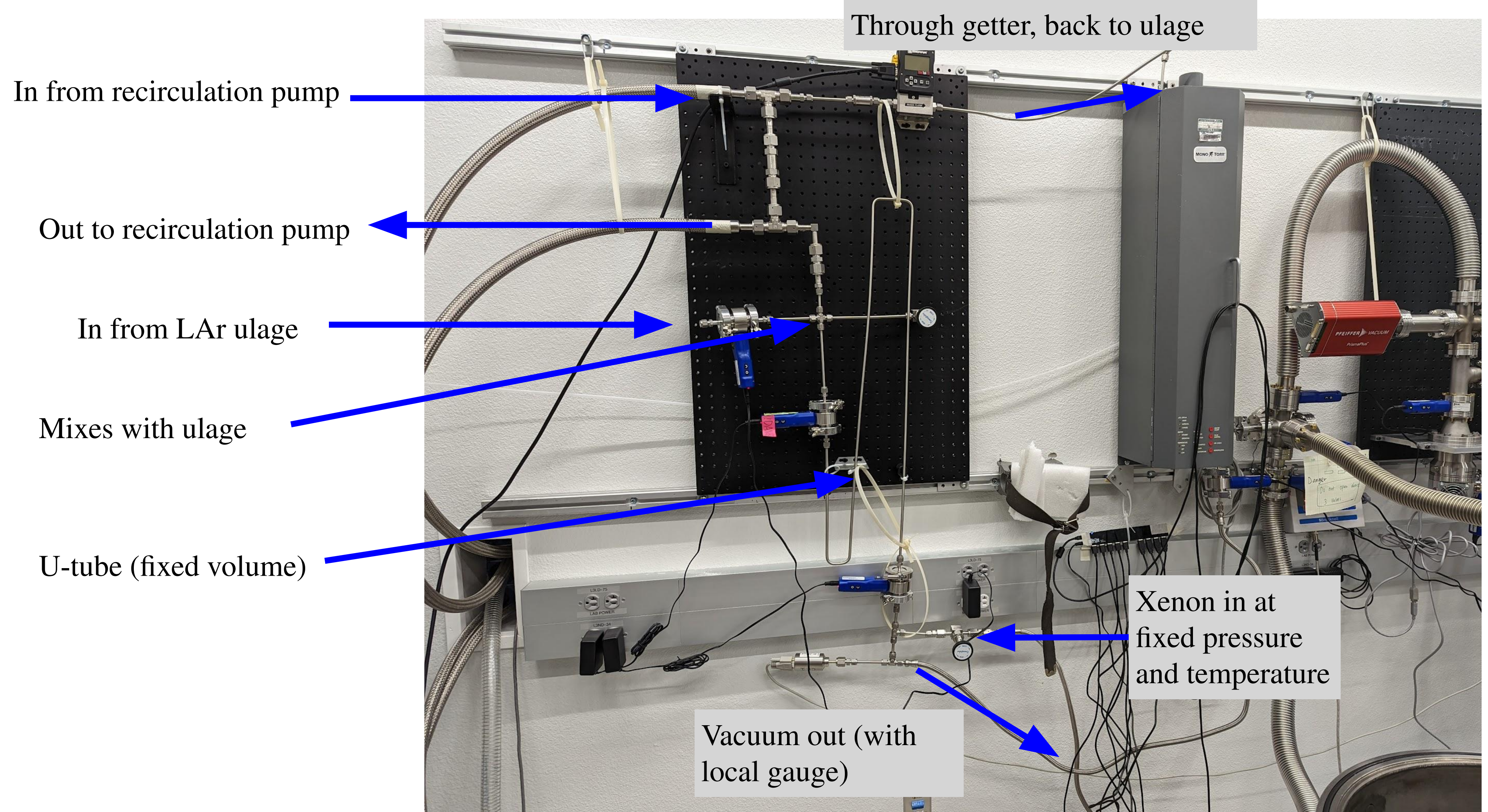
# New Setup

- One Hamamatsu R11410 (sensitive to 175 nm, but not 128 nm).
- Triggered  $^{241}\text{Am}$  source aka Llama (LEGEND LAr monitoring system from TUM).
- Three levels of 3 SiPM (Hamamatsu 13370) detectors sensitive to both 128 nm and 175 nm.
- One SiPM covered with Suprasil glass to block 128 nm light.



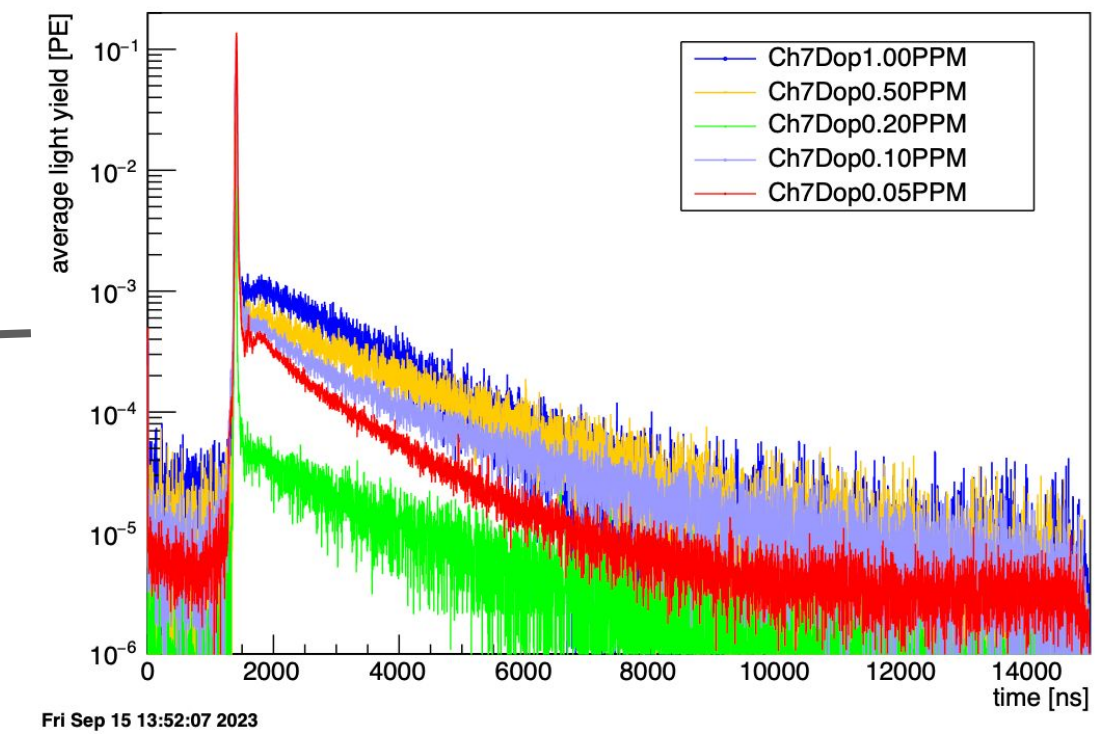
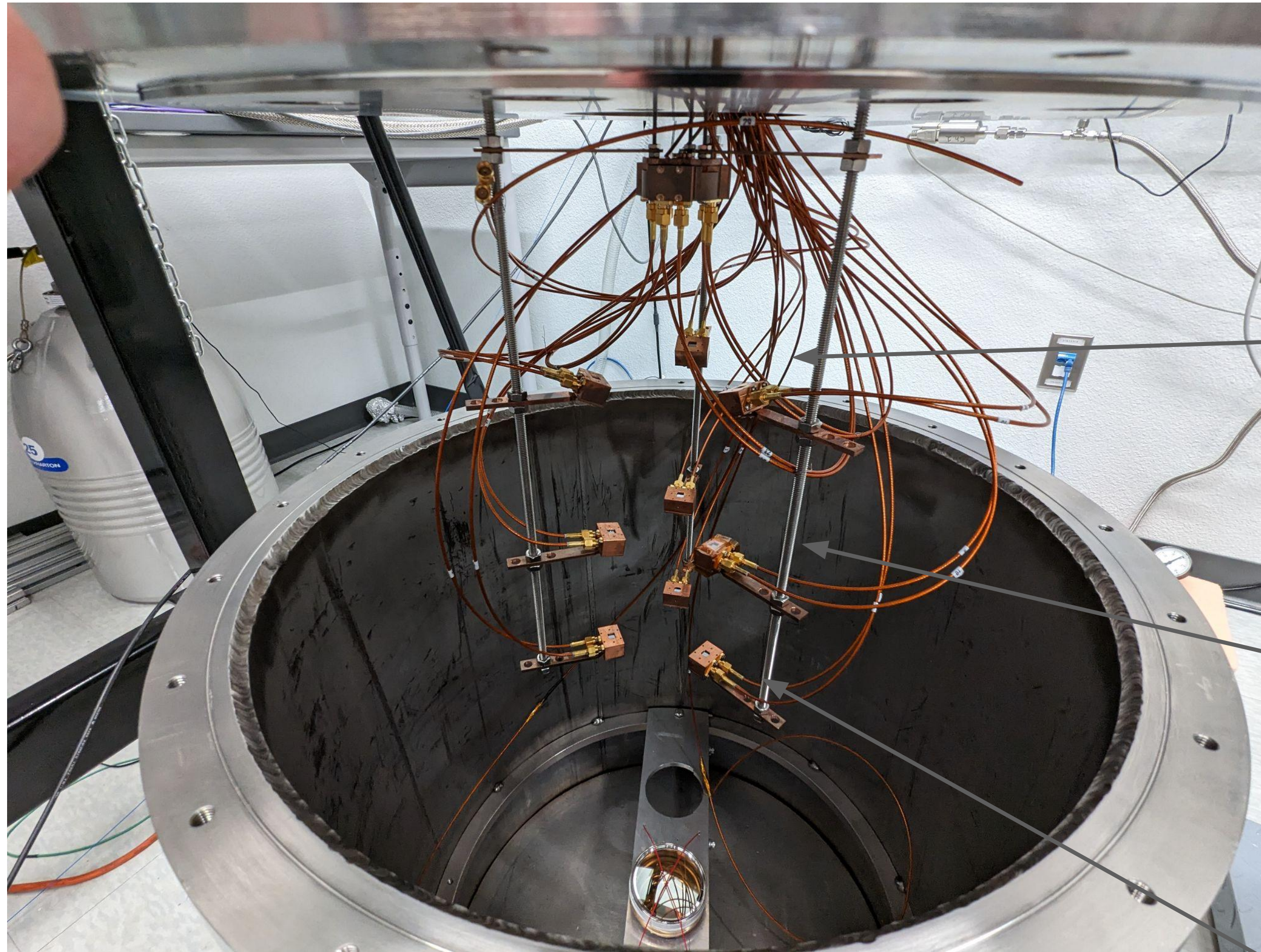


# Doping Procedure



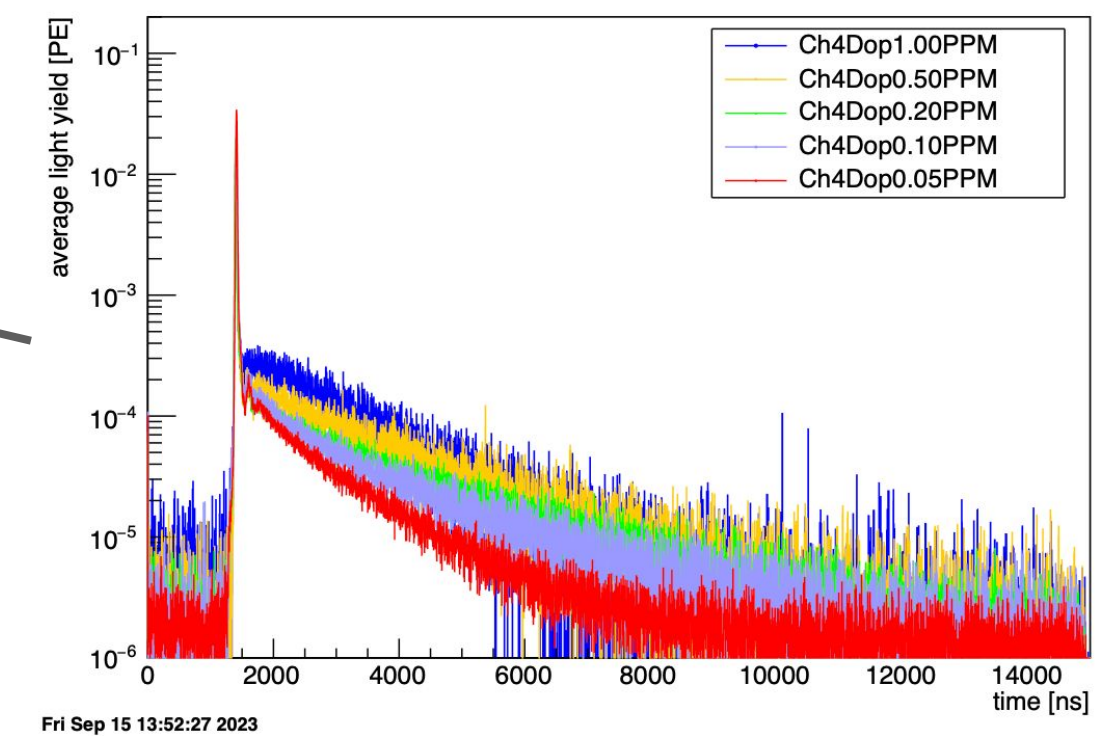


# Preliminary Data



Undoped 0.01 - 0.07 ppm

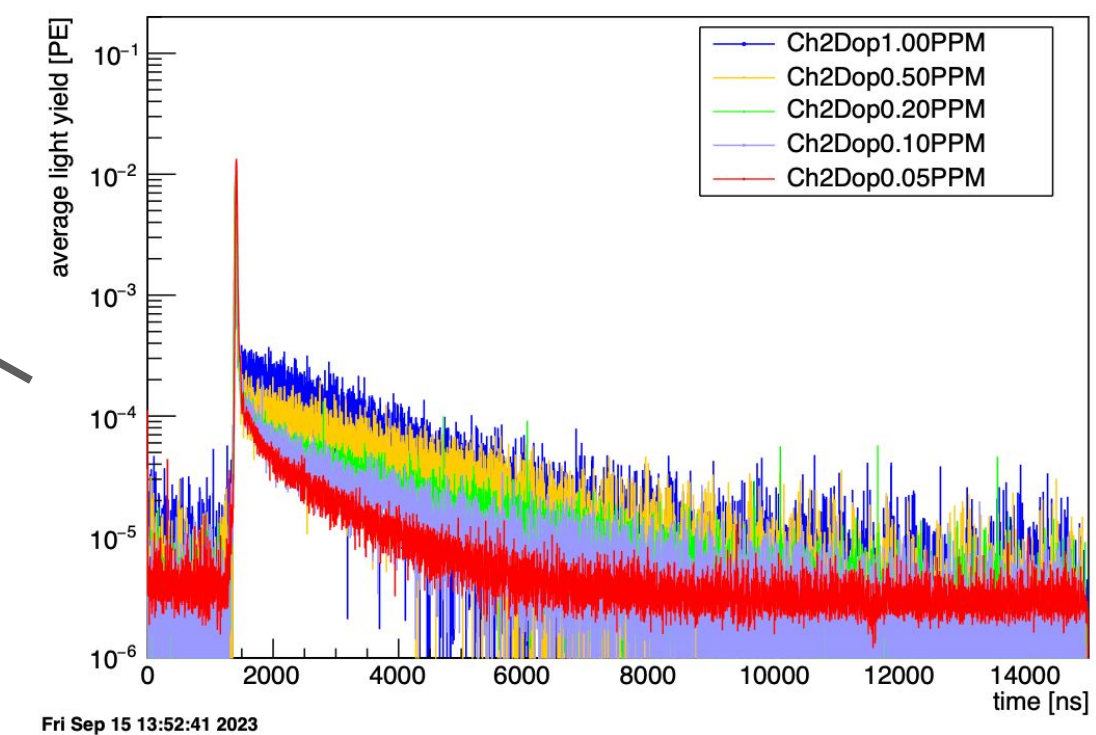
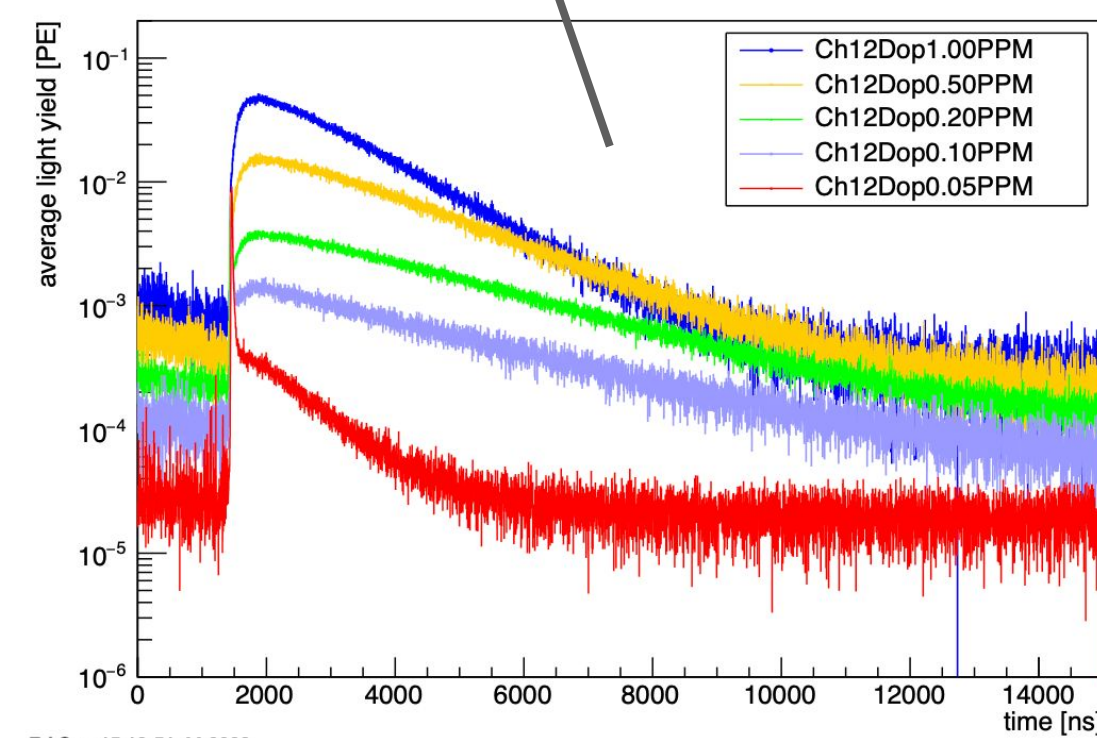
Doped 0.1 ppm



Doped 0.2 ppm

Doped 0.5 ppm

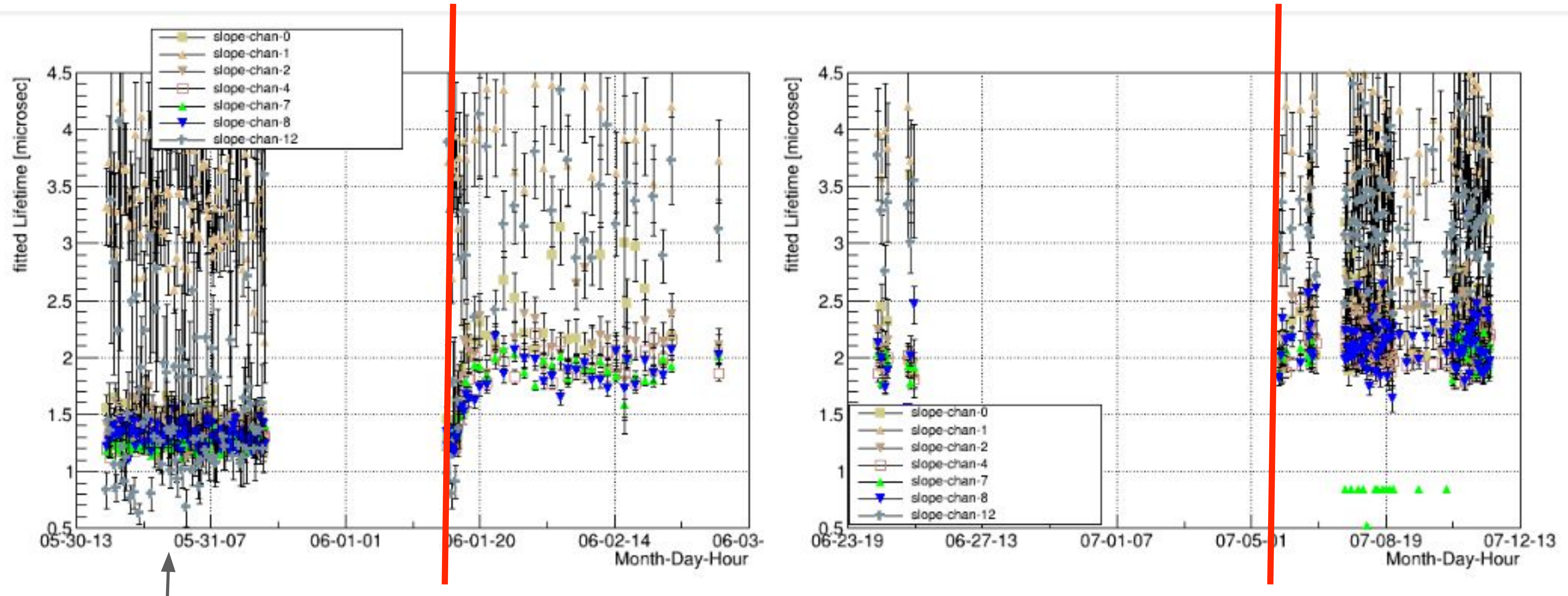
Doped 1.0 ppm





# Preliminary Xe-Doping Data

Slopes from  $\sim 1.5 \mu\text{s}$  after singlet to  $5 \mu\text{s}$



initial xenon concentration in ulage measured to be  $0.01 \pm 0.07$  ppm

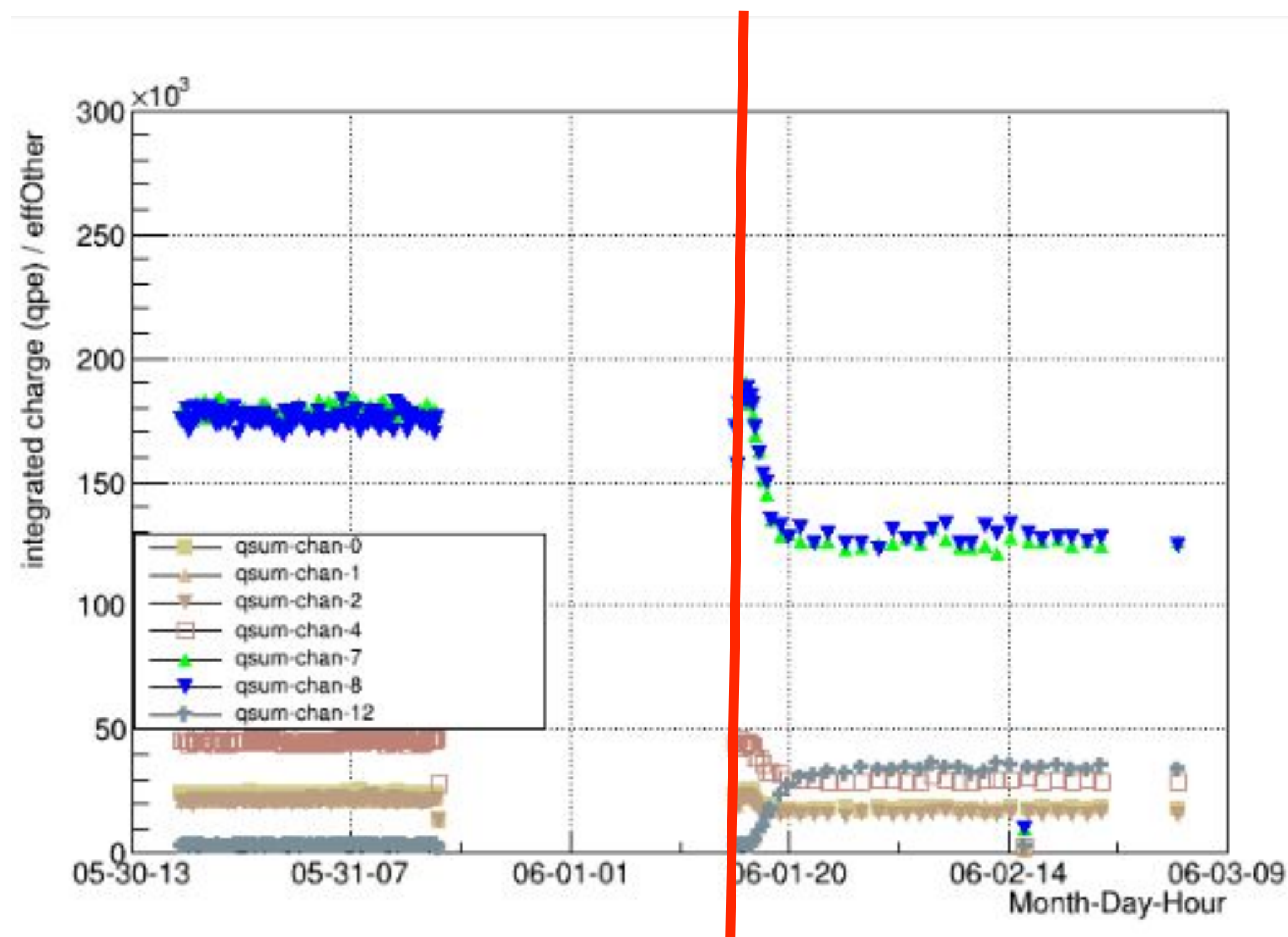
0.1 ppm (added to initial)

0.2 ppm total added

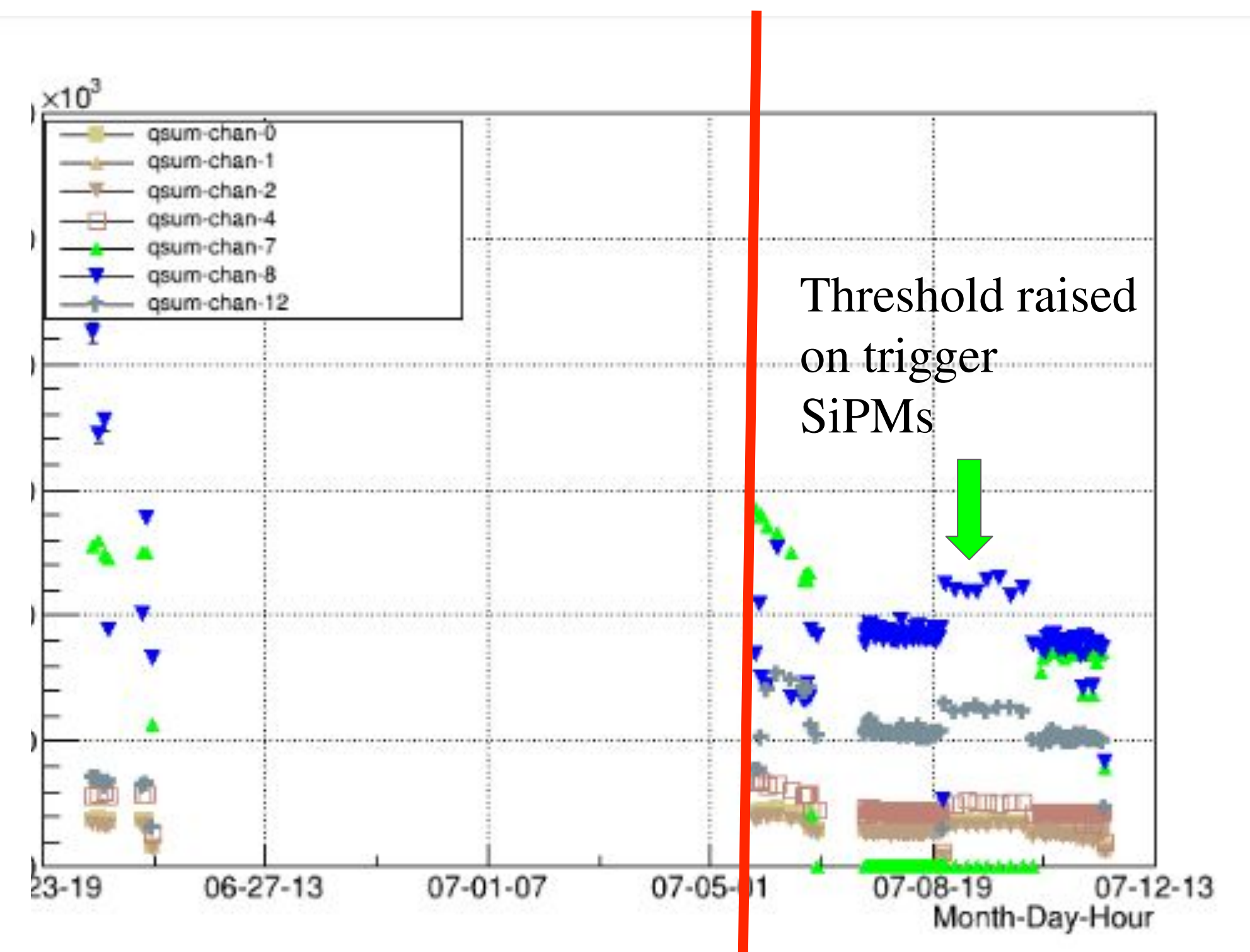


# Preliminary Xe-Doping Data

Integrated charge from  $\sim 1.5 \mu\text{s}$  after singlet to  $5 \mu\text{s}$



0.1 ppm (added to initial)

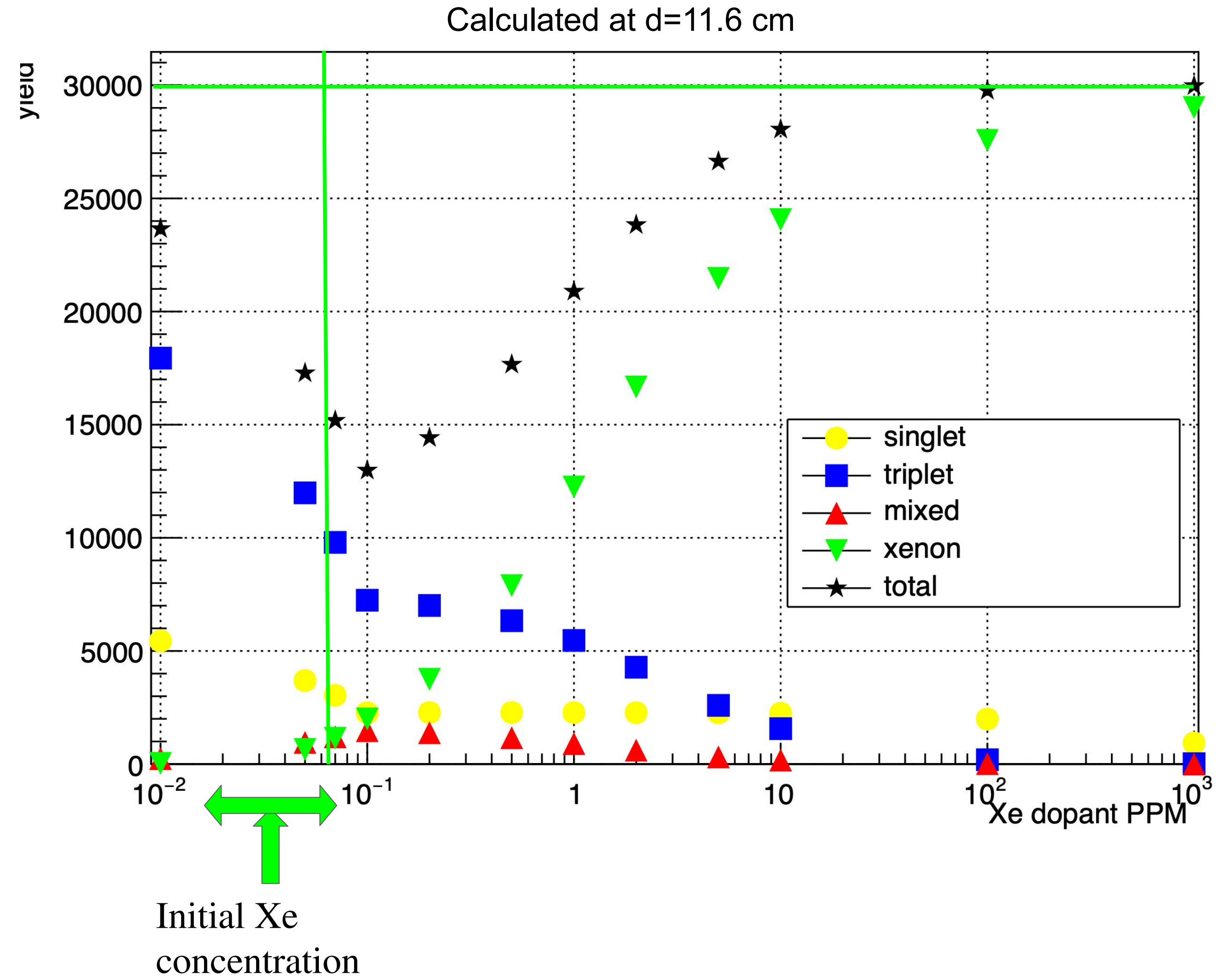


0.2 ppm total added



# Model Prediction (NOT yet tuned to data)

- At doping levels greater than 0.1, we take  $A$  to be constant (as in our fits), but at very low PPM, the assumption of constant absorption is no longer valid, so we use values of  $A$  from the previous fit.
- From the model we can understand the increase in light given the initial concentration of Xe in the LAr.
- Notice that there is still very strong reduction in light at 0.1 ppm Xe
- Even at 0.01 ppm, the yield is still only 80% of original (number of excitons).
- At 10 ppm Xe, light is recovered to >90% of original.



# Conclusions

- Xenon absorption of Argon 128 nm scintillation light is complicated and has a significant effect on the time spectra of emission and light yields even at very low concentrations.
- Quenching of long-lived states can account for increased light yields at Xe doping levels above 1 ppm.
- Our upcoming runs should narrow the parameter space for lifetimes and quenching times, and we can examine the effect on the fast/slow ratio for PID.
- Can also add  $N_2$ , etc. to the model...
- Thank you for staying to the end!

# Backup



# Scintillation light in LAr

There have been many papers and much work done to characterize and understand the amount and time distribution of scintillation light in liquid argon...

*Proc. Roy. Soc. Lond. A.* **317**, 113–131 (1970)  
Printed in Great Britain

## Experimental evidence in liquid argon

BY B. RAZ  
Department of Chemistry, UNM

(Communicated by Sir Nevill Martin Martin)

In connexion with studies of the electron-impact ionization of liquid argon, it was found whether there exist exciton states in liquid argon. A spectroscopic study of liquid argon showed that experimental evidence was obtained for Wannier exciton states in the excited states of the liquid. The absorption spectra of the doped liquid argon were measured. The following experimental results are reported:  
(a) In the Xe/Ar liquid two absorption bands were observed at 141 nm (8.80 eV) and at 123 nm (10.1 eV).  
(b) In the Xe/Kr liquid three absorption bands were observed at 125.5 nm (9.89 eV) and 129 nm (9.6 eV).  
(c) The absorption spectra of the doped liquid argon are shown in Fig. 1.

16v3 [physics.ins-det] 21 Sep 2019

Novosibirsk State University - Novosibirsk, 630090, Russia

## Fast component re-emission in Xe-doped liquid argon

D. Akimov,<sup>a,b</sup> V. Belov,<sup>a,b</sup> A. Konovalov,<sup>a,b,c</sup> A. Kumpan,<sup>b</sup> O. Razuvaeva,<sup>b</sup> D. Rudik,<sup>a,b,1</sup> and G. Simakov<sup>a,b,c</sup>

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**ABSTRACT:** We present the first direct experimental confirmation of the fast component re-emission in liquid argon (LAr) doped with xenon (Xe). This effect was studied at various Xe concentrations up to ~3000 ppm. The rate constant of energy transfer for the fast component was found to be independent of the Xe concentration. It was shown that LAr doped with a high concentration of Xe without TPB has a better Pulse Shape Discrimination (PSD) efficiency than pure LAr or Xe-doped LAr with TPB. The stability of the fast component in the mixture was tested for the first time at high Xe concentration for long continuous runtime.

## Attenuation of vacuum ultraviolet light in pure and xenon-doped liquid argon - an approach to an assignment of the near-infrared emission from the mixture

A. Neumeier<sup>1</sup>, T. Dandl<sup>2</sup>, A. Himpsl<sup>2</sup>, L. Oberauer<sup>1</sup>, W. Potzel<sup>1</sup>, S. Schönert<sup>1</sup>, and A. Ulrich<sup>2a</sup>

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<sup>2</sup> Technische Universität München, Physik-Department E12, James-Franck-Str. 1, D-85748 Garching, Germany

Published in EPL (2015)

**Abstract.** Results of transmission experiments of vacuum ultraviolet light through a 11.6 cm long cell filled with pure and xenon-doped liquid argon are described. Pure liquid argon shows no attenuation down to the experimental short-wavelength cut-off at 118 nm. Based on a conservative approach, a lower limit of 1.10 m for the attenuation length of its own scintillation light could be derived. Adding xenon to liquid argon at concentrations on the order of parts per million leads to strong xenon-related absorption features which are used for a tentative assignment of the recently found near-infrared emission observed in electron-beam excited liquid argon-xenon mixtures. Two of the three absorption features can be explained by perturbed xenon transitions and the third one by a trapped exciton (Wannier-Mott) impurity state. A calibration curve connecting the equivalent width of the absorption line at 140 nm with xenon concentration is provided.

## Emission

Work supported by FAPESP  
São Paulo, Brazil

Dark Matter experiments utilize liquid argon (LAr) as a scintillator. Liquid argon emits light at 127 nm and with a decay time of the lowest lying ground state. A model for the emission of the ground state is presented. The dependence of the intensity of the emission on the abundance of the ground state is studied experimentally.

## Table-top setup for investigating the scintillation properties of liquid argon

T. Heindl<sup>1</sup>, T. Dandl<sup>1</sup>, A. Fedenev<sup>2</sup>, M. Hofmann<sup>3</sup>, R. Krücken<sup>1</sup>, L. Oberauer<sup>3</sup>, W. Potzel<sup>3</sup>, J. Wieser<sup>4</sup>, and A. Ulrich<sup>1</sup>

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<sup>3</sup>Excitech GmbH, Planckstr. 1, 64291 Darmstadt, Germany

<sup>4</sup>Excitech GmbH, Branterei 33, 26419 Schortens, Germany

## IST (2011)

Light and temporal light emission properties of liquid argon have been studied in the context of rare-gas detectors for detecting Dark Matter particles in astronomy. A table-top setup for continuous and pulsed low energy electron beam excitation is used to stimulate light emission. The emission spectrum from 110 to 1000 nm in wavelength is covered by the detection system with a time resolution of 1 ns.

doi: 10.1209/0295-5075/109/12001

## Intense vacuum ultraviolet and infrared scintillation of liquid Ar-Xe mixtures

A. NEUMEIER<sup>1</sup>, T. DANDL<sup>2</sup>, T. HEINDL<sup>2</sup>, A. HIMPSL<sup>2</sup>, L. OBERAUER<sup>1</sup>, W. POTZEL<sup>1</sup>, S. ROTH<sup>1</sup>, S. SCHÖNERT<sup>1</sup>, J. WIESER<sup>3</sup> and A. ULRICH<sup>2(a)</sup>

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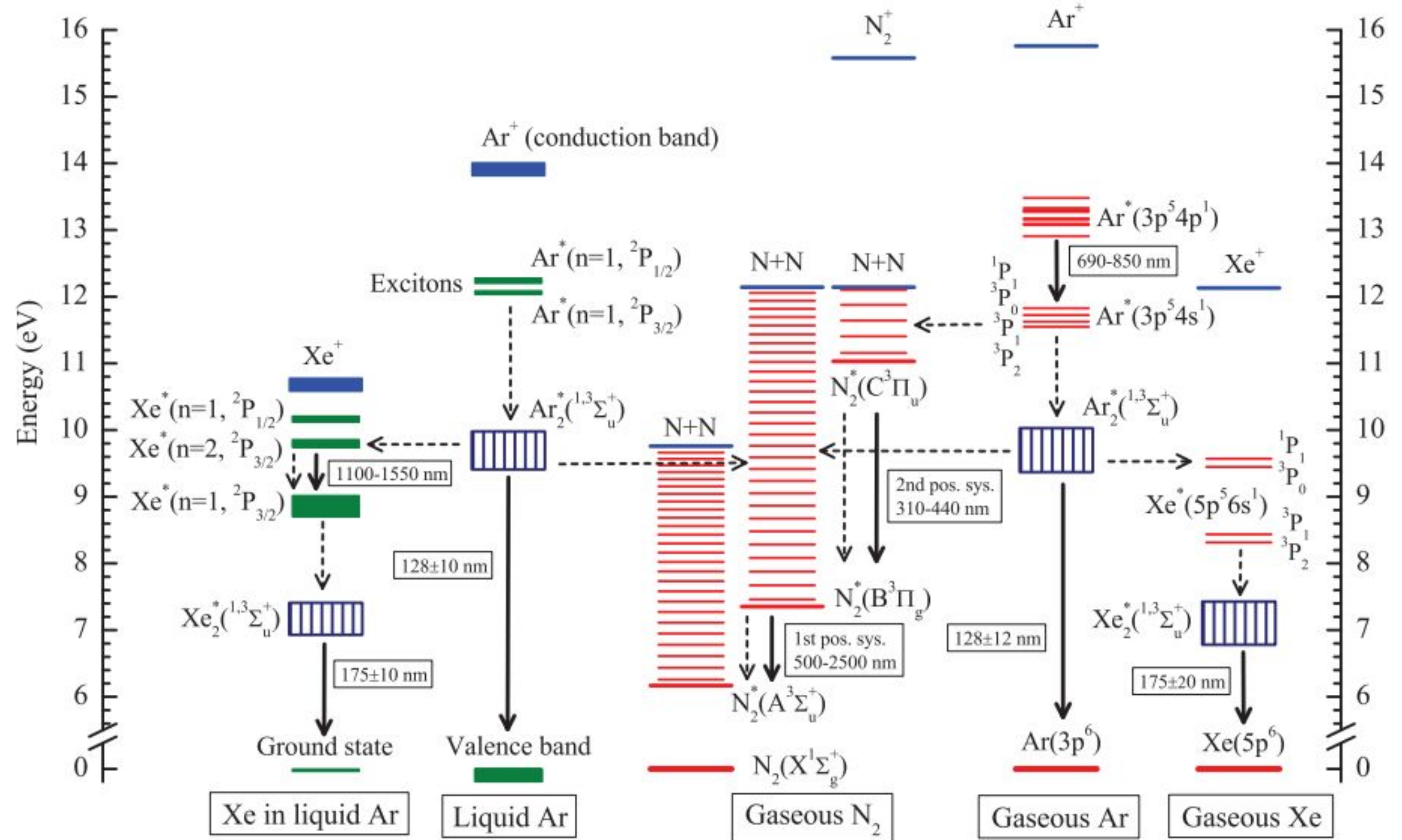
Scintillation light in xenon-doped liquid argon is described in the present paper. The emission spectrum in liquid argon from 0.1 ppm xenon concentration to 1000 ppm is presented. The emission spectrum of liquid argon from the second excimer continuum of argon (~174 nm) is observed by recording optical emission spectra at a xenon concentration of ~10 ppm for which a peak wavelength of 1.17  $\mu\text{m}$  with  $13000 \pm 4000$  counts per second is found. The corresponding value for the VUV excimer continuum of xenon is determined to be 1.17  $\mu\text{m}$ . Under these excitation conditions pure liquid argon shows a peak wavelength of 1.17  $\mu\text{m}$ . The electron energy deposited at a peak wavelength of 1.17  $\mu\text{m}$  is determined to be 1.17 eV. The emission spectrum for the 10 ppm Ar-Xe liquid mixture shows a peak wavelength of 1.17  $\mu\text{m}$ . UV emission spectra from xenon-doped liquid argon from 0.1 ppm to 1000 ppm are also shown. The emission spectrum is presented at well-defined wavelength positions in the present paper.



# Starting Point

A. Buzulutskov

Not working in an (academic) vacuum, but until now, no one has put all the pieces together...





# Plus Nitrogen

- The addition of Nitrogen involves a complicated cascade of reactions, some of which emit in the IR and UV...
- BUT, decay times are very long (from 10  $\mu\text{s}$  to several ms), hence collisional quenching dominates and one can take excitation energy transfer to Nitrogen excimers as non-photonic.
- So, one could just add another loss term with a time constant equal to the diffusion time times the nitrogen concentration.